Claims

1. A process for making rare earth doped optical fibre using stable dispersions (sol) of RE coated silica nanoparticles and applying a thin coating of the said silica sol containing suitable dopants selected from Ge, Al, P etc.

2. A process for making rare earth doped optical fibre, said process comprising steps of:

   (a) obtaining rare earth oxide doped silica nanoparticles by sonochemical method;

   (b) preparing stable dispersions of the above RE containing powders in the desired proportion in presence of suitable dopants like Al^{3+}, Ge^{4+} etc. in a silica sol of $Si(OC_2H_5)_4$ under sonication.;

   (c) applying a thin coating of silica sol on the inner surface of high purity clear fused silica glass tubes by sol-gel dip coating technique;

   (d) drying the coated layer in air at 70° to 150°C;

   (e) mounting the tube on glass working lathe for processing by MCVD technique;

   (f) dehydrating the coated layer inside the tube at a temperature in the range of 800-1200 °C in presence of excess Cl_2 ;

   (g) sintering the coated layer in presence of a mixture of oxygen and helium in the temperature range of 1400 to 1750°C for formation of the core;

   (h) heating the tube gradually up to a temperature of 1900°C for further consolidation of the sintered layer;

   (i) collapsing the tube by usual method at a temperature in the range of 2000-2300 °C to obtain a preform;

   (j) overcladding the preform with silica tube, and

   (k) drawing fibres of standard dimensions from the preform by the conventional methods.

3. A process as claimed in claim 1 wherein, the RE oxide is selected from Eu_2O_3, Nd_2O_3, Tb_2O_3 and Er_2O_3 for preparation of the silica nanoparticles.

4. A process as claimed in claim 1 wherein, P_2O_5 and F doped synthetic cladding is deposited within a silica glass substrate tube prior to development of the coating by

known method like Modified Chemical Vapour Deposition (MCVD) process to obtain matched or depressed clad type structure in the preform.

5. A process as claimed in claim 1 wherein, the particle size of the RE coated SiO_2 powders ranges from 50 to 200 nm.
6. A process as claimed in claim 1 wherein, the composition in oxide mol% of SiO_2: Re_2O_3 in Re_2O_3 coated SiO_2 powders varies from 99.5 : 0.5 to 95 : 5.
7. A process as claimed in claim 1 wherein, the equivalent oxide mol% of SiO_2 in the dispersion varies from 98.5 to 90.5.
8. A process as claimed in claim 1 wherein, a silica sol prepared with $Si(OC_2H_5)_4$ was used as the diluent of the RE_2O_3 coated silica powder.
9. A process as claimed in claim 1 wherein, the equivalent oxide mol% of GeO_2 in the dispersion varies from 1.0 to 5.0 .
10. A process as claimed in claim 1 wherein, Ge^{4+} was added through $Ge(OC_2H_5)_4$ in the silica sol.
11. A process as claimed in claim 1 wherein, the equivalent oxide mol% of Al_2O_3 in the dispersion ranges from 0.5 to 4.0.
12. A process as claimed in claim 1 wherein, Al_2O_3 is provided to the solvent in the form of aluminium salts such as chlorides, nitrates or any other salt soluble in the solvent.
13. A process as claimed in claim 1 wherein, the solution of aluminium salt is prepared using a solvent selected from alcohol and water.
14. A process as claimed in claim 1 wherein, the oxide mol% of Er_2O_3 in the dispersion ranges from 0.01 to 0.60.
15. A process as claimed in claim 1 wherein, strong mineral acids used for preparing the sol for dispersion are selected from hydrochloric or nitric acid.
16. A process as claimed in claim 1 wherein, the alcohol selected is soluable in the dispersion system.
17. A process as claimed in claim 1 wherein, the alcohol is selected from the group comprising of methyl alcohol, ethyl alcohol, propan-1-ol, propan-2-ol, butan-1-ol and butan-2-ol.

18. A process as claimed in claim 1 wherein, pH of the dispersion ranges from 1 to 5.
19. A process as claimed in claim 1 wherein, viscosity of the dispersion varies from 1 to 10 mPa s.
20. A process as claimed in claim 1 wherein, sonication time of the dispersion ranges from 30 to 200 minutes.
21. A process as claimed in claim 1 wherein, settling time of the dispersion varies from 1 to 10 hours.
22. A process as claimed in claim 1 wherein, lifting speed of the tube from the dispersion ranges from 4 to 15 cm/minutes.
23. A process as claimed in claim 1 wherein, baking temperature of the coated tube varies from 70° to 150°C.
24. A process as claimed in claim 1 wherein, baking time of the coated tube ranges from 0.5 to 5 hours.
25. A process as claimed in claim 1 wherein, the core composition is selected from the group comprising of RE_2O_3 +SiO_2+GeO_2, RE_2O_3 +SiO_2+GeO_2 +Al_2O_3, RE_2O_3 +SiO_2+GeO_2 +Al_2O_3 +P_2O_5 and RE_2O_3 +SiO_2+GeO_2 + P_2O_5.
26. A process as claimed in claim 1 wherein, the temperature of the RE oxide containing core layer is increased in steps of 50 to 200°C during sintering depending on the composition and Al/RE concentration of the core layer.
27. A process as claimed in claim 1 wherein, the mixture of O_2 and He is in the range of 3:1 to 9:1 during sintering.
28. A process as claimed in claim 1 wherein, source of chlorine is CCl_4 where helium is used as carrier gas.
29. A process as claimed in claim 1 wherein, the proportion of Cl_2: O_2 during drying varies from 1.5 : 1 to 3.5 : 1.
30. A process as claimed in claim 1 wherein, the dehydration period lies between 1 to 2 hours.
31. A process as claimed in claim 1 wherein, the core layer is sintered in the presence of germania to facilitate germania incorporation and to obtain appropriate numerical aperture value.